

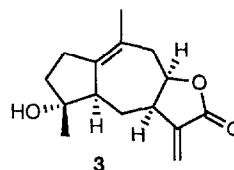
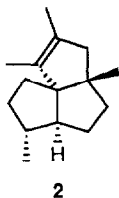
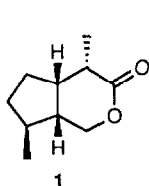
DIASTEREOSELECTIVE ZIRCONOCENE-PROMOTED BICYCLIZATION-CARBONYLATION OF  
ALLYLICALLY METHYL-SUBSTITUTED ENYNES.  
SYNTHESIS OF (+)-IRIDOMYRMECIN

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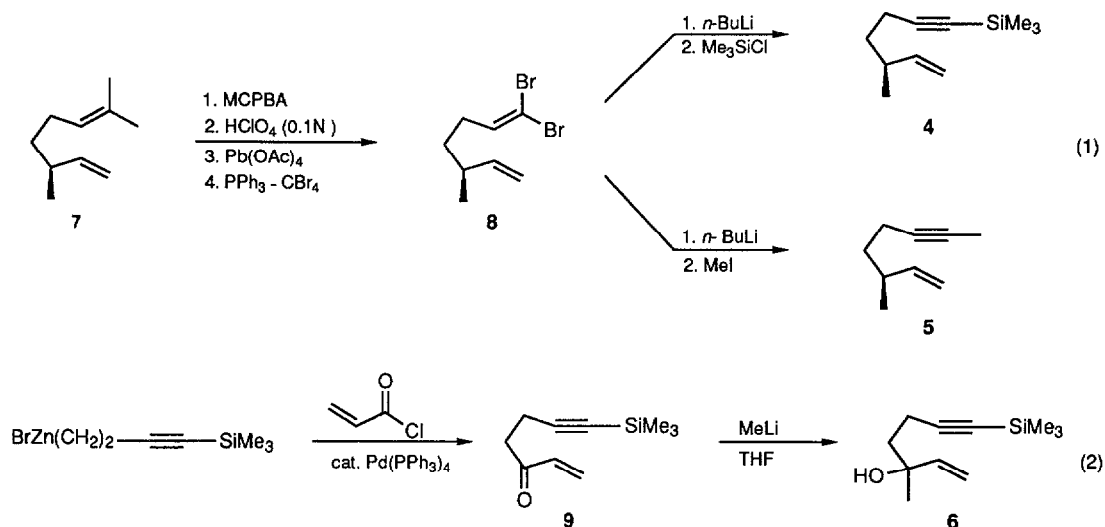
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**Summary:** Allylically methyl-substituted 1,6-heptynynes, such as 3-methyl-1,6-octenyne, can be diastereoselectively converted to the corresponding bicyclic ketones, such as (5*R*,6*S*)-2,6-dimethylbicyclo[3.3.0]oct-1-en-3-one readily convertible to (+)-iridomyrmecin, the observed diastereomeric excesses of the bicyclization reaction being >90%.

It has recently been demonstrated that the zirconocene-promoted bicyclization of allylically hydroxy-substituted dienes<sup>1</sup> and enynes<sup>2</sup> can be diastereoselective and that the corresponding bicyclic ketones,<sup>1a,2b</sup> can be stereoselectively prepared.<sup>3</sup> In view of a large number of natural products containing nonpolar substituents, especially Me, in place of OH, such as iridomyrmecin<sup>4</sup> (1), silphiperfol-6-ene<sup>5</sup> (2) and pseudoivalin<sup>6</sup> (3), that are potentially accessible via the bicyclization-carbonylation sequence,<sup>7</sup> we decided to investigate the effects of a Me group in the allylic position of enynes on the stereochemistry of the zirconocene-promoted bicyclization. This study was also prompted by our recent finding that the course of the zirconocene-promoted alkyl-alkene and alkene-alkene coupling via zirconacyclopentane formation is strongly influenced by even small nonpolar substituents, such as Me.<sup>8</sup>

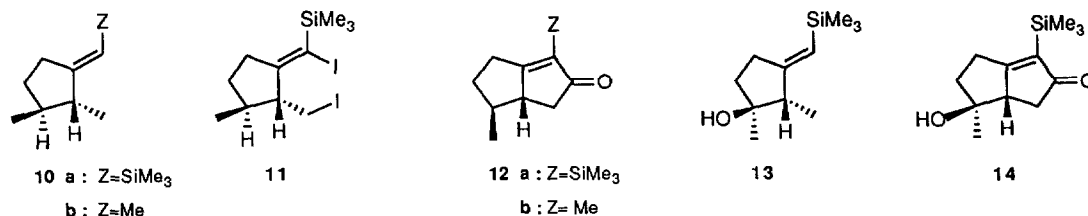


Three allylically Me-substituted 1,6-heptynynes 4-6 were prepared as follows. (+)-Citronellene (7),  $[\alpha]_D^{20} +10 \pm 1^\circ$ , available from Fluka was converted in 46% yield to 8 via epoxidation with *m*-ClC<sub>6</sub>H<sub>4</sub>COOH, oxidation with Pb(OAc)<sub>4</sub>,<sup>9</sup> and one carbon homologation with CBr<sub>4</sub> and PPh<sub>3</sub> in CH<sub>2</sub>Cl<sub>2</sub> at 0 °C.<sup>10</sup> Treatment of 8 with *n*-BuLi (2.5 equiv) in THF followed by quenching with Me<sub>3</sub>SiCl gave a 91% yield of 4, while quenching with MeCl provided 5 in 78% yield (eq. 1). The palladium-catalyzed reaction<sup>11</sup> of Me<sub>3</sub>SiC≡C(CH<sub>2</sub>)<sub>2</sub>ZnBr<sup>12</sup> with acryloyl chloride gave 9 in 63% yield. Its reaction with MeLi afforded 6 in 72% yield (eq. 2).



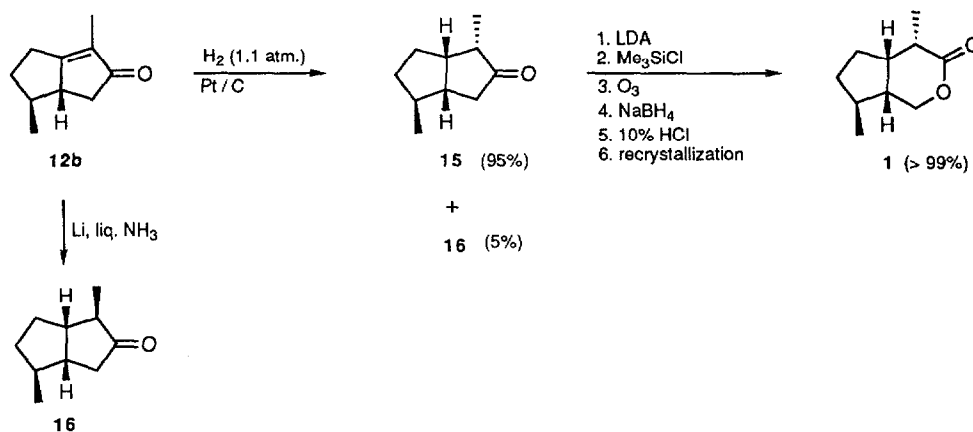
The enynes **4-6** were treated with a  $\text{ZrCp}_2$  reagent generated in situ by the reaction of  $\text{Cp}_2\text{ZrCl}_2$  with 2 equiv of  $n\text{-BuLi}$ .<sup>7,13</sup> Without isolation and/or characterization, the zirconabicyclic products were subjected to further reactions. Protonolysis (3N HCl), iodolysis ( $\text{I}_2$  in THF), and carbonylation followed by quenching with  $\text{I}_2$ <sup>8</sup> of the zirconabicyclic derived from **4** produced **10a** (80%), **11** (55%), and **12a** (41%) in the yields indicated in parentheses. The diastereomeric excess in each case was >96% by  $^{13}\text{C}$  NMR spectroscopy. Similarly, **5** was converted to **10b** (91%) and **12b** (73%) via protonolysis and carbonylation-iodolysis, respectively. These reactions were >90% d.e. When the bicyclization reaction of **5** was carried out at 25 °C only for 3 h, protonolysis led to the formation of a 4:1 mixture of the two possible diastereomers. On the other hand, protonolysis after 18 h at 25 °C led to the >90% d.e. figure reported above. We judge that the reaction is reversible as previously indicated for a similar enyne<sup>7</sup> and that the fraction of the minor isomer under the kinetically controlled conditions is larger than that under the thermodynamically equilibrated conditions. If so, equilibration in the case of **4** must be considerably faster than that of **5**, since the >96% d.e. figure was obtained only after 3 h at 25°C. Quite unexpected was the >96% d.e. figure that the bicyclization reaction of **6** displayed. Protonolysis and carbonylation-iodolysis of the bicyclization mixture gave **13** and **14** in 75 and 35% yields, respectively. Although the factors influencing the diastereochemistry of the reaction are not clear at this time, the predominant formation of the *exo*-OH isomer suggests that the OH group converted to a metallated derivative containing Li and/or Zr under the bicyclization conditions must exert a considerably greater steric demand than Me.

The stereochemical assignments of the bicyclic ketones are based on their  $^1\text{H}$  NMR spectra including  $^1\text{H}$  2D NOESY NMR spectra. The identity of **12b**,  $[\alpha]_{\text{D}}^{20} -78^\circ$  (*c* 10.2,  $\text{CHCl}_3$ ), was further confirmed by comparison of its spectra with those reported for its enantiomer,  $[\alpha]_{\text{D}}^{20} +77.8^\circ$  (*c* 9.42,  $\text{CHCl}_3$ ), which has been prepared as an intermediate for the synthesis of **2**.<sup>5b</sup>



To demonstrate the applicability of the zirconocene-promoted bicyclization-carbonylation protocol to natural products synthesis, conversion of **12b** into (+)-iridomyrmecin<sup>4</sup> (**1**) was achieved as shown in Scheme I. Catalytic hydrogenation of **12b** (>98% d.e.) at 1.1 atm over 1% Pt/C in MeOH for 4 h at 25 °C gave **15** as a ~95% isomerically pure compound in 84% yield, the only isomer detectable by <sup>13</sup>C NMR spectroscopy being the 2-Me epimer (**16**). The use of Pd/C as a catalyst led only to a 3:1 mixture of **15** and **16**. An essentially 100% pure sample of **16** was readily obtainable via reduction of **12b** with Li in liquid NH<sub>3</sub>. One-pot conversion of **15** into **1** was achieved as in Scheme I, following a sequence reported in the literature.<sup>4b</sup> After a short-path column chromatography (silica gel, 4:1 hexane/EtOAc), a 97:3 mixture of **1** and its epimer, i.e., isoiridomyrmecin, was obtained in 54% overall yield based on **15**. Recrystallization from hexane provided a 46% overall yield of a ≥99% pure sample of **1**:<sup>4</sup> mp 60.5-61.0 °C (lit.<sup>4a</sup> mp 60.5-61.0 °C); [α]<sub>D</sub><sup>23</sup> +208° (lit.<sup>4a</sup> [α]<sub>D</sub> +210°); <sup>1</sup>H NMR (CDCl<sub>3</sub>, Me<sub>4</sub>Si) δ 0.95-1.25 (m, 2H), 1.06 (d, *J* = 6 Hz, 3H), 1.15 (d, *J* = 6.5 Hz, 3H), 1.7-1.9 (m, 4H), 2.5-2.8 (m, 2H), 4.17 (d, *J* = 12 Hz, 1H), 4.28 (dd, *J* = 12 and 3 Hz, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, Me<sub>4</sub>Si) δ 12.76, 18.41, 29.89, 34.26, 37.38, 38.02, 41.27, 45.60, 68.13, 176.78; IR (Nujol) 1760 cm<sup>-1</sup>.

### Scheme I



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